**Q1.Which of the following complexes obeys to EAN:**

[Ni(CO)4] , [Fe2(CO)9] , [HMn(CO)3(C4H6)] , [Fe(CN)6]3- , [H2W2(CO)8]

**Q2. Give a reasonable explanation for the following:**

**1-**Most metal nitrosyls and carbonyls are electronically stable.

**2-**Cp2Mn is readily reduced by alkaline metal to Mn1-.

**3-** Hexacarbonyl vanadium(0) can ‘nt dimmer at RT for the sake of Sidgwick

**4-**Most cluster and sandwich metal complexes possess 18 e’s rule.

**GOOD LUCK**

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