## 3-4 Particle in a box

Let's now try to investigate a free particle moves in a potential distribution as shown below;

$$
V(x)=\left\{\begin{array}{ll|l|l}
0 & 0 \leq x \leq a \\
\infty & 0 \geq x \geq a
\end{array}\right) \psi=0, \begin{array}{ll}
E=? & \psi=? \\
& V(x)=\infty
\end{array}
$$

It is seen that for such a system the particle is restricted to move in the confined region $0 \leq x \leq a$ due to the infinite potential value that prevents the particle to penetrate the barrier at $x=0$ and $x=a$. This system looks like a molecule in a gas cylinder that can't escape out due to the solid walls of the cylinder. In addition, this system is similar to a free electron in metals.
H.W: Prove mathematically that $\psi=0$ in the regions $0 \geq x \geq a$ and state the physical meaning for that.

Anyway Schrodinger equation in the region under consideration is given by;

$$
\begin{equation*}
\frac{d^{2} \psi(x)}{d x^{2}}+k^{2} \psi(x)=0 \tag{3-23}
\end{equation*}
$$

Keeping in mind that $k^{2}=\frac{2 m E}{\hbar^{2}}$, the solution of this equation is;

$$
\begin{equation*}
\psi(x)=A e^{i k x}+B e^{-i k x} \tag{3-24}
\end{equation*}
$$

The constants $A$ and $B$ can found as before with aid of the boundary condition as in below. Indeed, at $x=0$ the wave function in (3-24) vanishes, i.e. $\psi(x=0)=0$. So, $A+B=0$ and hence $A=-B$, therefore;

$$
\psi(x)=A\left(e^{i k x}-e^{-i k x}\right)
$$

Using Euler's method last equation becomes; $e^{ \pm i \theta}=\cos \theta \pm i \sin \theta$

$$
\begin{aligned}
& \psi(x)=A(\cos k x+i \sin k x-\cos k x+i \sin k x) \\
& \psi(x)=2 i A \sin k x \\
& \psi(x)=C \sin k x
\end{aligned}
$$

Where $C=2 i A$, and similarly at $x=a$ the wave function is also vanishes, i.e. $\psi(x=a)=0$, therefore;

$$
\psi(x=a)=C \sin k a=0
$$

This requires that; $\sin k a=0$, which leads to $k a=n \pi$ and thus;

$$
k=\frac{n \pi}{a}
$$

Consequently, the energy eigen values and eigen wave functions of a particle moving in a potential box becomes respectively as follows;

$$
E_{n}=\frac{\pi^{2} \hbar^{2} n^{2}}{2 m a^{2}} \text { and } \psi_{n}(x)=C \sin (n \pi x / a)
$$

It is obvious that the energy of this system is quantize because the particle is confine in a limited region by means of the potential. Accordingly, we may draw the following;

$$
\begin{array}{ll}
n=4 & E_{4}=16 E_{1} \\
n=3 & E_{3}=9 E_{1} \\
n=2 & E_{2}=4 E_{1} \\
n=1 & E_{1}=\frac{\pi^{2} \hbar^{2}}{2 m a^{2}}
\end{array}
$$

Indeed, the energy of ground state, $E_{1}=\frac{\pi^{2} \hbar^{2}}{2 m a^{2}}$ call Zero Point Energy. Concerning with the wave function $\left(\psi_{n}(x)=C \sin (n \pi x / a)\right)$ the normalization constant $C$ can found as follows;

$$
\begin{aligned}
& \int \psi_{n}^{*}(x) \psi_{n}(x) d x=1 \\
& \int C^{*} \sin (n \pi x / a) C \sin (n \pi x / a) d x=1 \\
& C^{2} \int_{0}^{a} \sin ^{2}(n \pi x / a) d x=1
\end{aligned}
$$

Using the identity; $\sin ^{2} \theta=\frac{1}{2}(1-\cos 2 \theta)$, we get;

$$
\begin{aligned}
& \frac{1}{2} C^{2}\left\{\int_{0}^{a} d x-\int_{0}^{a} \cos (2 n \pi x / a) d x\right\}=1 \\
& \frac{1}{2} C^{2}\left\{\left.x\right|_{0} ^{a}-\left.\left(\frac{a}{2 n \pi}\right) \sin (2 n \pi x / a)\right|_{0} ^{a}\right\}=1 \\
& \frac{1}{2} C^{2}\{(a-0)-(a / 2 n \pi)(0-0)\}=1 \\
& \frac{1}{2} C^{2} a=1 \quad C=\sqrt{\frac{2}{a}} \\
& \therefore \quad \psi_{n}(x)=\sqrt{\frac{2}{a}} \sin (n \pi x / a)
\end{aligned}
$$

The figure below shows the first three eigen wave functions $\psi_{1}, \psi_{2}, \psi_{3}$.


### 3.5 Particle in potential box in three dimensions

The same procedure used in last section repeats here but in three dimensions $x, y$ and $z$ instead of one. So, the particle now moves freely in a box of dimensions $a, b$ and $c$ as shown in figure below.


The Schrödinger equation for this system must written as follows;

$$
\begin{equation*}
\frac{\partial^{2} \psi(x, y, z)}{\partial x^{2}}+\frac{\partial^{2} \psi(x, y, z)}{\partial y^{2}} \frac{\partial^{2} \psi(x, y, z)}{\partial z^{2}}+k^{2} \psi(x, y, z)=0 \tag{3-25}
\end{equation*}
$$

Keeping in mind that; $k^{2}=\frac{2 m E}{\hbar^{2}}$. However, to solve this equation the method of separating of variables must be use. Accordingly, we may write;

$$
\begin{equation*}
\psi(x, y, z)=\psi(x) \cdot \psi(y) \cdot \psi(z) \tag{3-26}
\end{equation*}
$$

The substitution of equation (3-26) into equation (2-25) and dividing the resultant expression on equation (3-26) leads to;

$$
\begin{equation*}
\frac{1}{\psi(x)} \frac{d^{2} \psi(x)}{d x^{2}}+\frac{1}{\psi(y)} \frac{d^{2} \psi(y)}{d y^{2}}+\frac{1}{\psi(z)} \frac{d^{2} \psi(z)}{d z^{2}}+k^{2}=0 \tag{3-27a}
\end{equation*}
$$

Or;

$$
\begin{equation*}
\frac{1}{\psi(x)} \frac{d^{2} \psi(x)}{d x^{2}}+k_{x}^{2}+\frac{1}{\psi(y)} \frac{d^{2} \psi(y)}{d y^{2}}+k_{y}^{2}+\frac{1}{\psi(z)} \frac{d^{2} \psi(z)}{d z^{2}}+k_{z}^{2}=0 \tag{3-27b}
\end{equation*}
$$

Which is equivalent to;

$$
\begin{align*}
& \frac{d^{2} \psi(x)}{d x^{2}}+k_{x}^{2} \psi(x)=0 \\
& \frac{d^{2} \psi(y)}{d y^{2}}+k_{y}^{2} \psi(y)=0  \tag{3-27c}\\
& \frac{d^{2} \psi(z)}{d z^{2}}+k_{z}^{2} \psi(z)=0
\end{align*}
$$

The solution of each of thses three equations is as shown below respectively;

$$
\begin{align*}
& \psi(x)=A_{x} e^{i k_{x}}+B_{x} e^{-i k_{x}} \\
& \psi(y)=A_{y} e^{i k_{y}}+B_{y} e^{-i k_{y}}  \tag{3-28}\\
& \psi(x)=A_{z} e^{i k_{x}}+B_{z} e^{-i k_{z}}
\end{align*}
$$

By using the boundary conditions $[\psi(x=0, x=a)=0, \psi(y=0, y=b)=0, \psi(z=0, z=c)$
$=0]$, we can get;

$$
\begin{align*}
\psi(x) & =C_{x} \operatorname{sink}_{x} x \\
\psi(y) & =C_{y} \sin k_{y} y  \tag{3-29}\\
\psi(z) & =C_{z} \operatorname{sink}_{z} z
\end{align*}
$$

Therefore the wave function of this system is as follows;

$$
\begin{equation*}
\psi(x, y, z)=C_{x} \sin \left(k_{x} x\right) \cdot C_{y} \sin \left(k_{y} y\right) \cdot C_{z} \sin \left(k_{z} z\right) \tag{3-30}
\end{equation*}
$$

Remember that; $k_{x}=n_{x} \pi / a, k_{y}=n_{y} \pi / b$ and $k_{z}=n_{z} \pi / c$ one get;

$$
\begin{equation*}
\psi(x, y, z)=C \sin \left(\frac{n_{x} \pi}{a} x\right) \cdot \sin \left(\frac{n_{y} \pi}{b} y\right) \cdot \sin \left(\frac{n_{z} \pi}{c} z\right) \tag{3-31}
\end{equation*}
$$

H.W: Show that; $C=\sqrt{\frac{8}{a b c}}$

Concerning with energy eigen values we have that; $E=\frac{p^{2}}{2 m}=\frac{\hbar^{2} k^{2}}{2 m}$, so;

$$
E=\frac{\hbar^{2}}{2 m}\left(k_{x}^{2}+k_{y}^{2}+k_{z}^{2}\right)
$$

The substituation of thevalues of $k_{x}=n_{x} \pi / a, k_{y}=n_{y} \pi / b$ and $k_{z}=n_{z} \pi / c$ leads to;

$$
\begin{equation*}
E=\frac{\hbar^{2} \pi^{2}}{2 m}\left(\frac{n_{x}^{2}}{a^{2}}+\frac{n_{y}^{2}}{b^{2}}+\frac{n_{z}^{2}}{c^{2}}\right) \tag{3-32}
\end{equation*}
$$

For the case when $a=b=c$ last formula becomes;

$$
\begin{equation*}
E=\frac{\hbar^{2} \pi}{2 m a^{2}}\left(n_{x}^{2}+n_{y}^{2}+n_{z}^{2}\right) \tag{3-33}
\end{equation*}
$$

Or equvilently;

$$
\begin{equation*}
E=E_{1}\left(n_{x}^{2}+n_{y}^{2}+n_{z}^{2}\right)=E_{1} n^{2} \tag{3-34}
\end{equation*}
$$

Equation (3-34) represent the the allowable energies eigen values, however it indicates that all of the states characterized by $\left(n_{x}, n_{y}, n_{z}\right)$ that gives the same value of $N$ have the same value of energy. Indeed, this is directly shows to the degeneracy where there are more than one wave function describe the same state, see the table below.

| Degeneracy | Combination of $\left(n_{x}, n_{y}, n_{z}\right)$ | Energy |
| :---: | :---: | :---: |
| 1 | $(1,1,1)$ | $3 E_{1}$ |
| 3 | $(1,1,2)(1,2,1)(2,1,1)$ | $6 E_{1}$ |
| 3 | $(1,2,2)(2,1,2)(2,2,1)$ | $9 E_{1}$ |
| 3 | $(3,1,1)(1,3,1)(1,1,3)$ | $11 E_{1}$ |
| 1 | $(2,2,2)$ | $12 E_{1}$ |
| 6 | $(1,2,3)(3,2,1)(2,3,1)(1,3,2)(2,1,3)(3,1,2)$ | $14 E_{1}$ |

