

Dr.Nadia Mohammed

Atmospheric chemistry

Regional Air Pollution

(...1950s...)

Climate Change

(...1990s...)

**Atmospheric
CHEMISTRY**

Acid rain

(1960s...)

**Stratospheric
Ozone depletion**

(1970s...)

- ➔ **Factors controlling the natural composition of the atmosphere**
- ➔ **Effects of human activities in changing atmospheric composition**



1948 – Noontime (Donora, PA)

Urban smog around the world today



New Delhi



Mexico City



Seoul



Beijing



Los Angeles

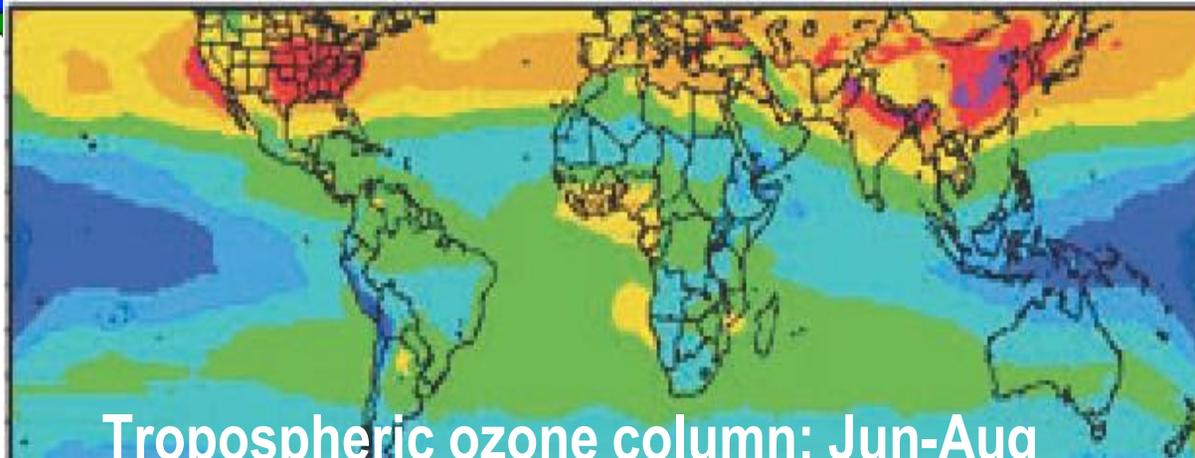
Urban/Regional air pollution



160 million people

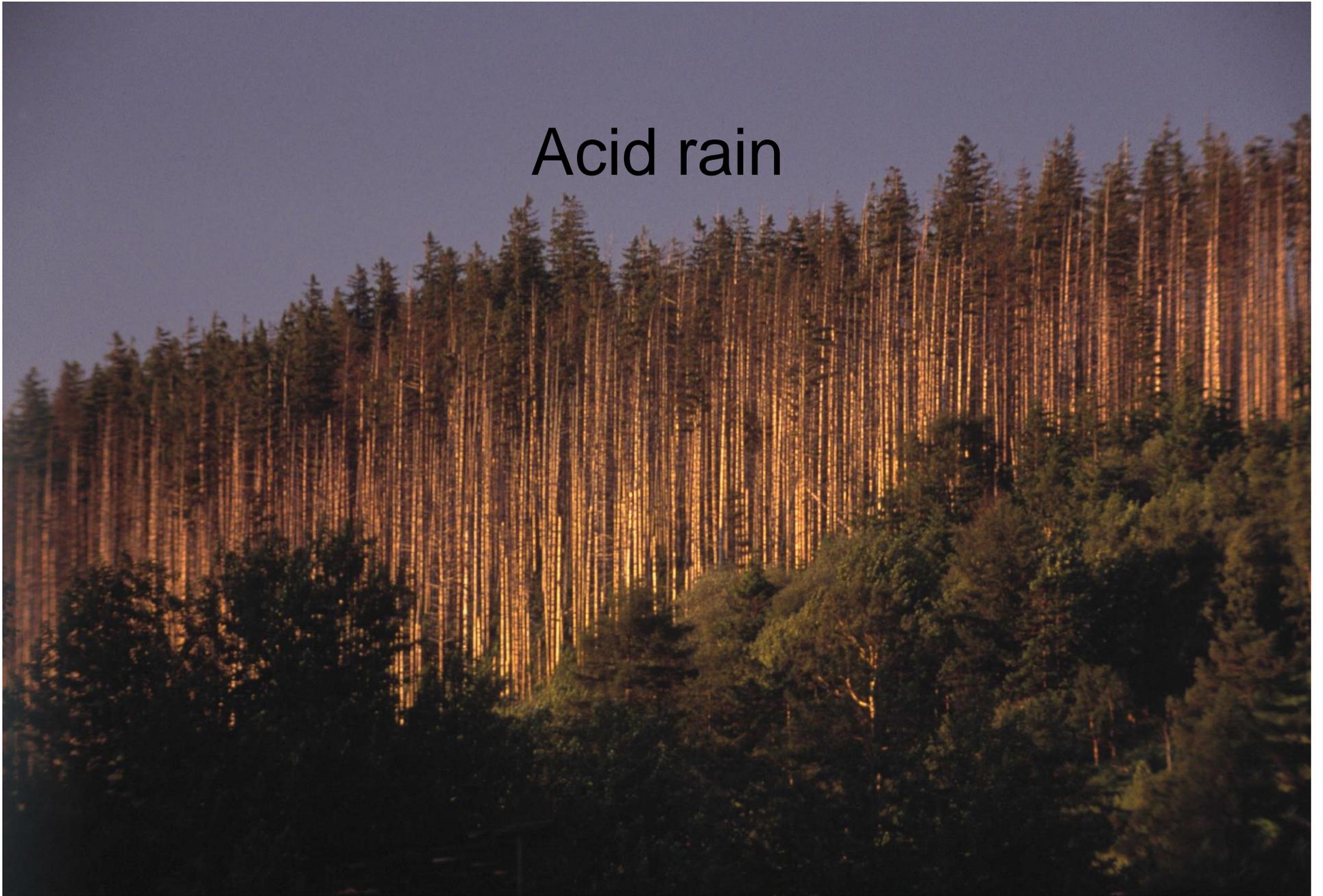
4/2011

nattainment for 5 NAAQS Pollutants
nattainment for 4 NAAQS Pollutants
nattainment for 3 NAAQS Pollutants
nattainment for 2 NAAQS Pollutants
nattainment for 1 NAAQS Pollutant



Tropospheric ozone column: Jun-Aug

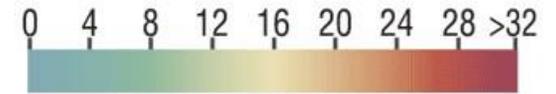
Acid rain



Acidified Forest, Oberwiesenthal, Germany (1991)

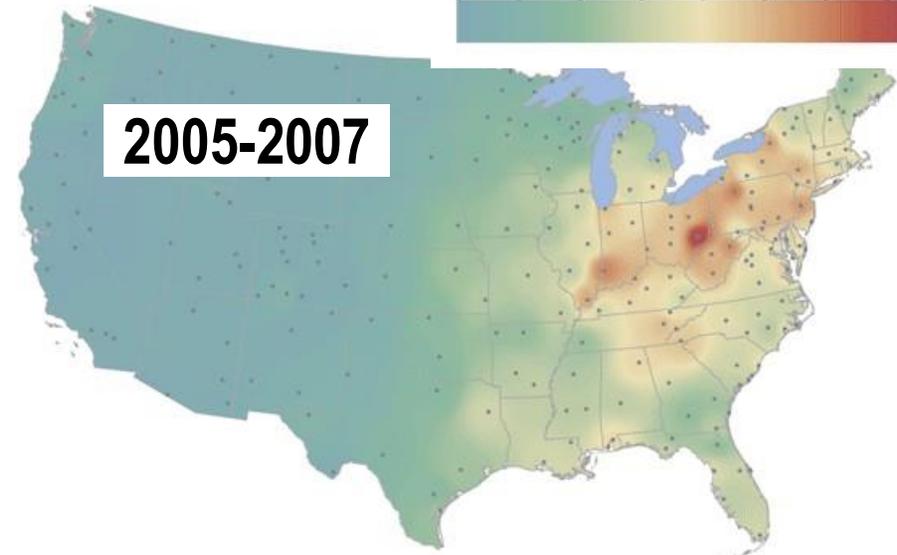
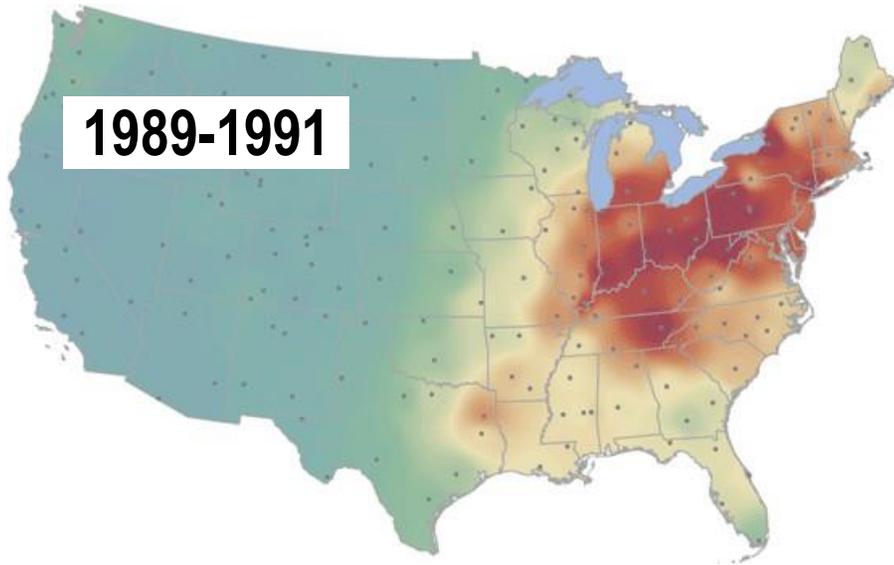
Acid rain

Wet SO_4^{2-} deposition
(kilograms per hectare):



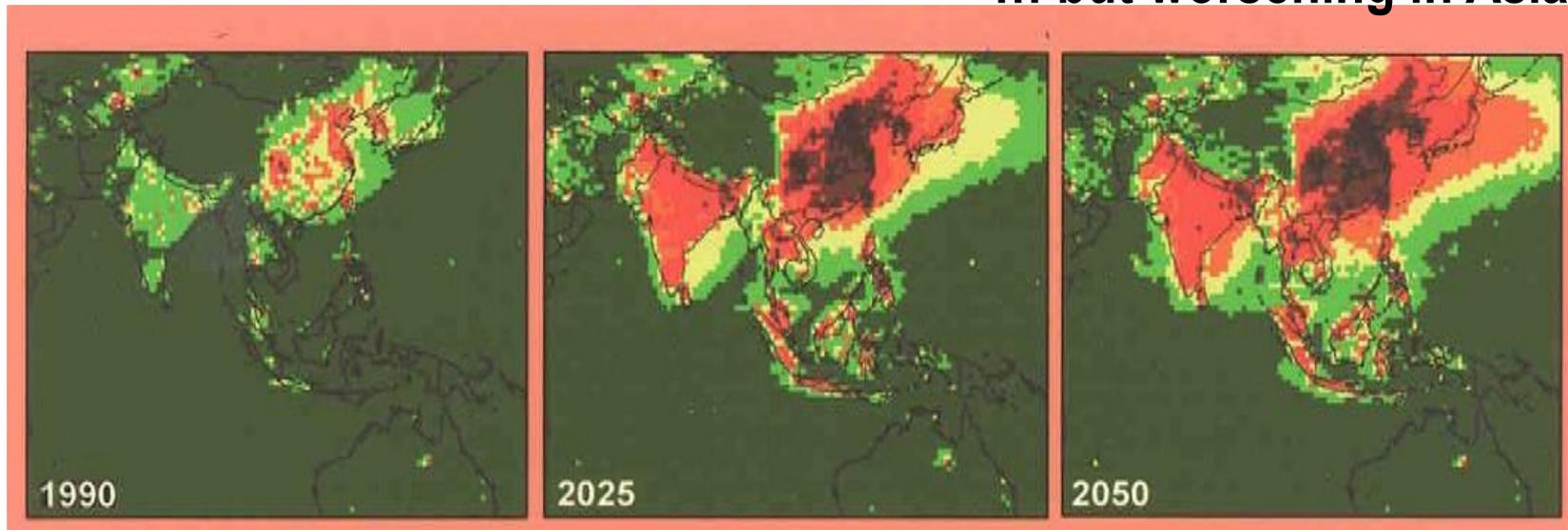
1989-1991

2005-2007



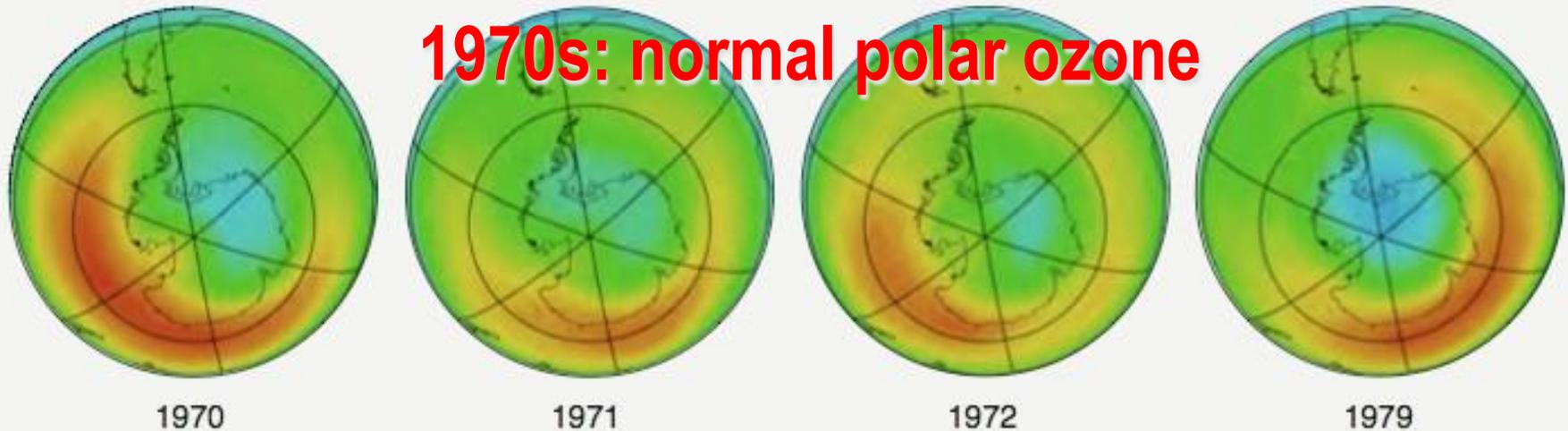
Wet sulfate (SO_4^{2-}) deposition

Getting better in the U.S. ...
... but worsening in Asia

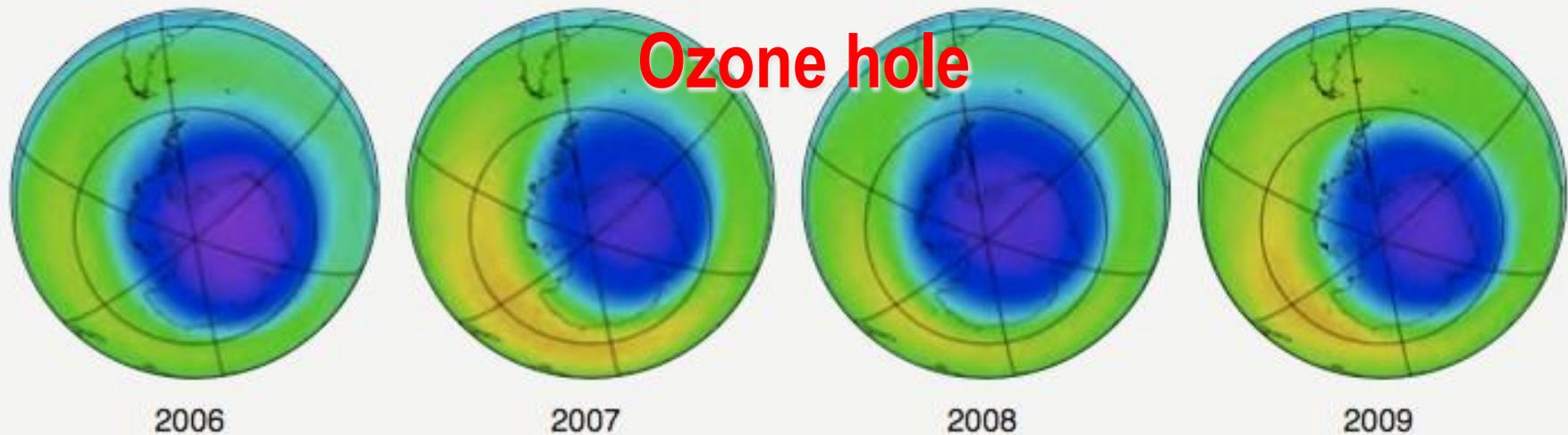


Stratospheric ozone depletion

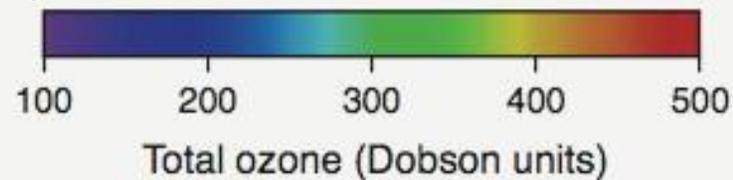
1970s: normal polar ozone



Ozone hole

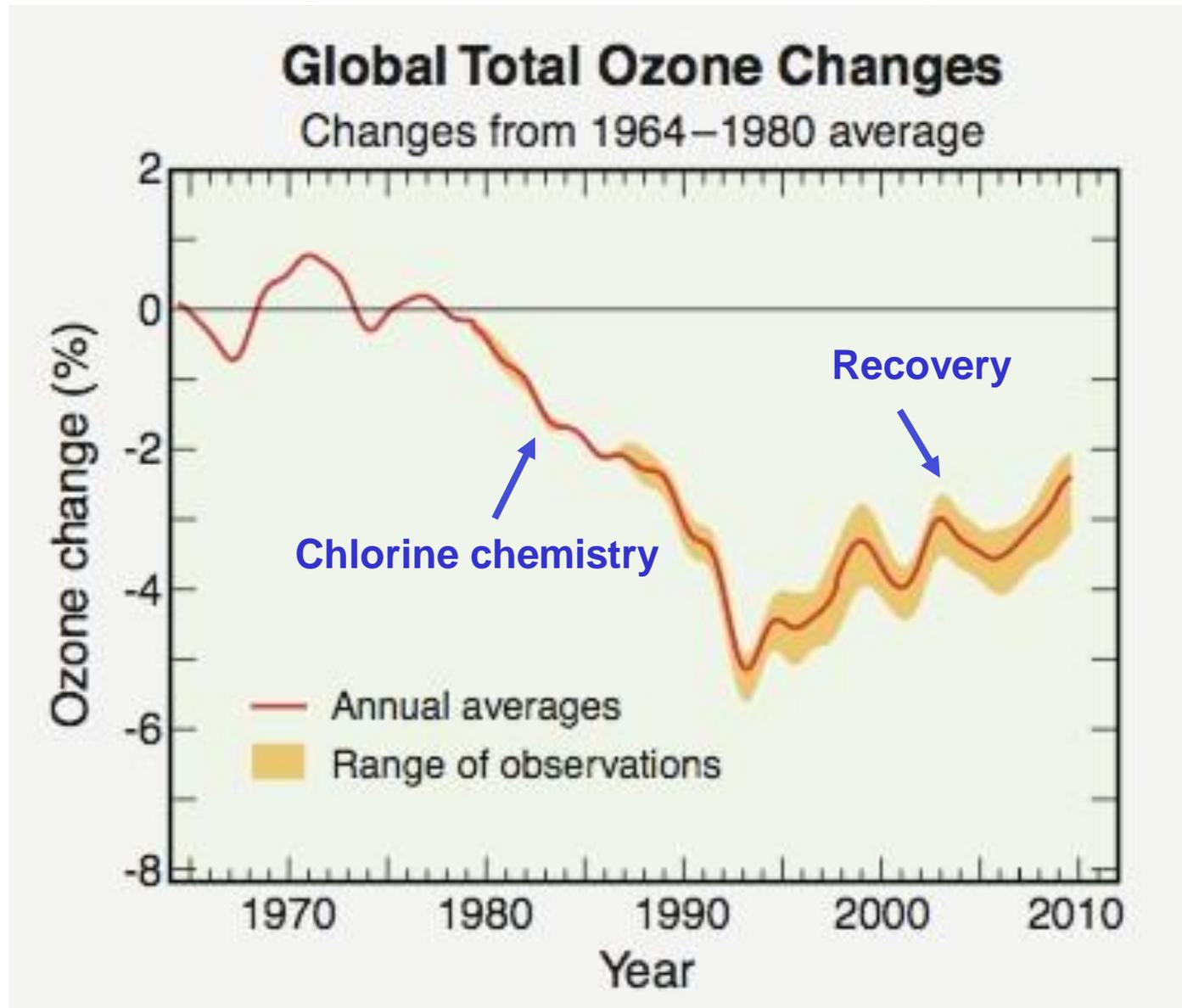


October monthly averages



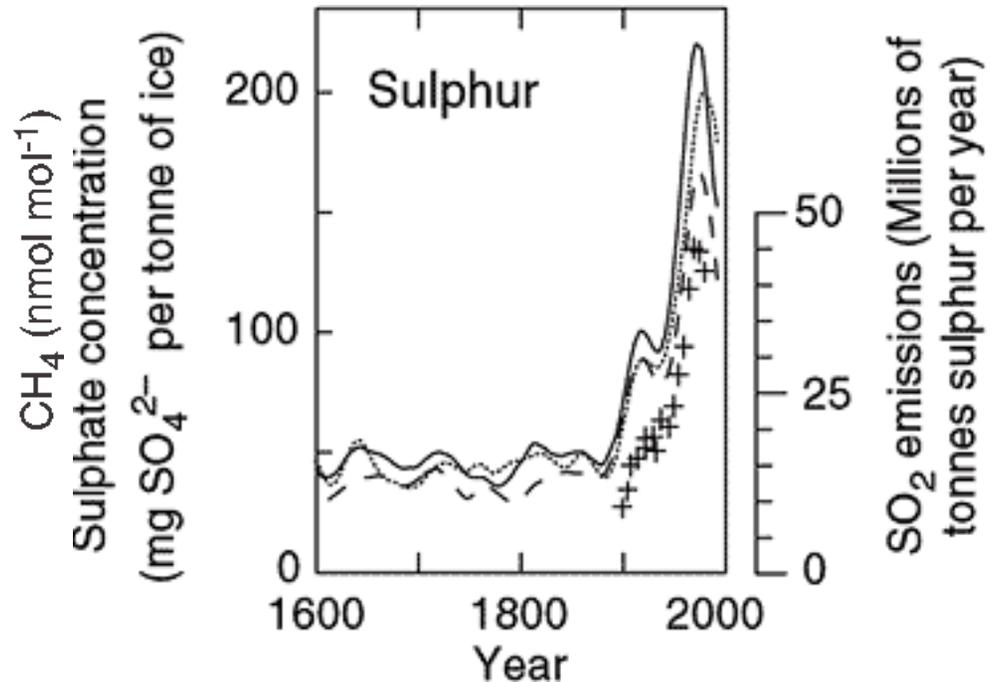
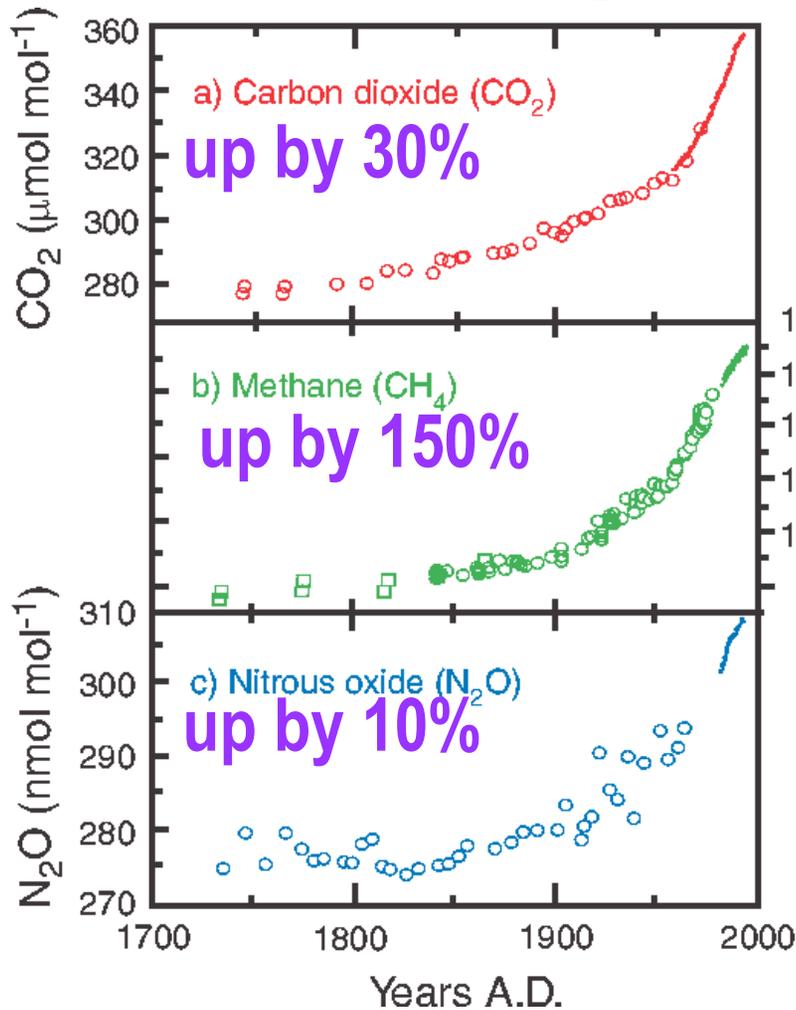
[WMO, 2010]

Stratospheric ozone depletion



➔ Downward trends in ozone column on a global scale

Climate change



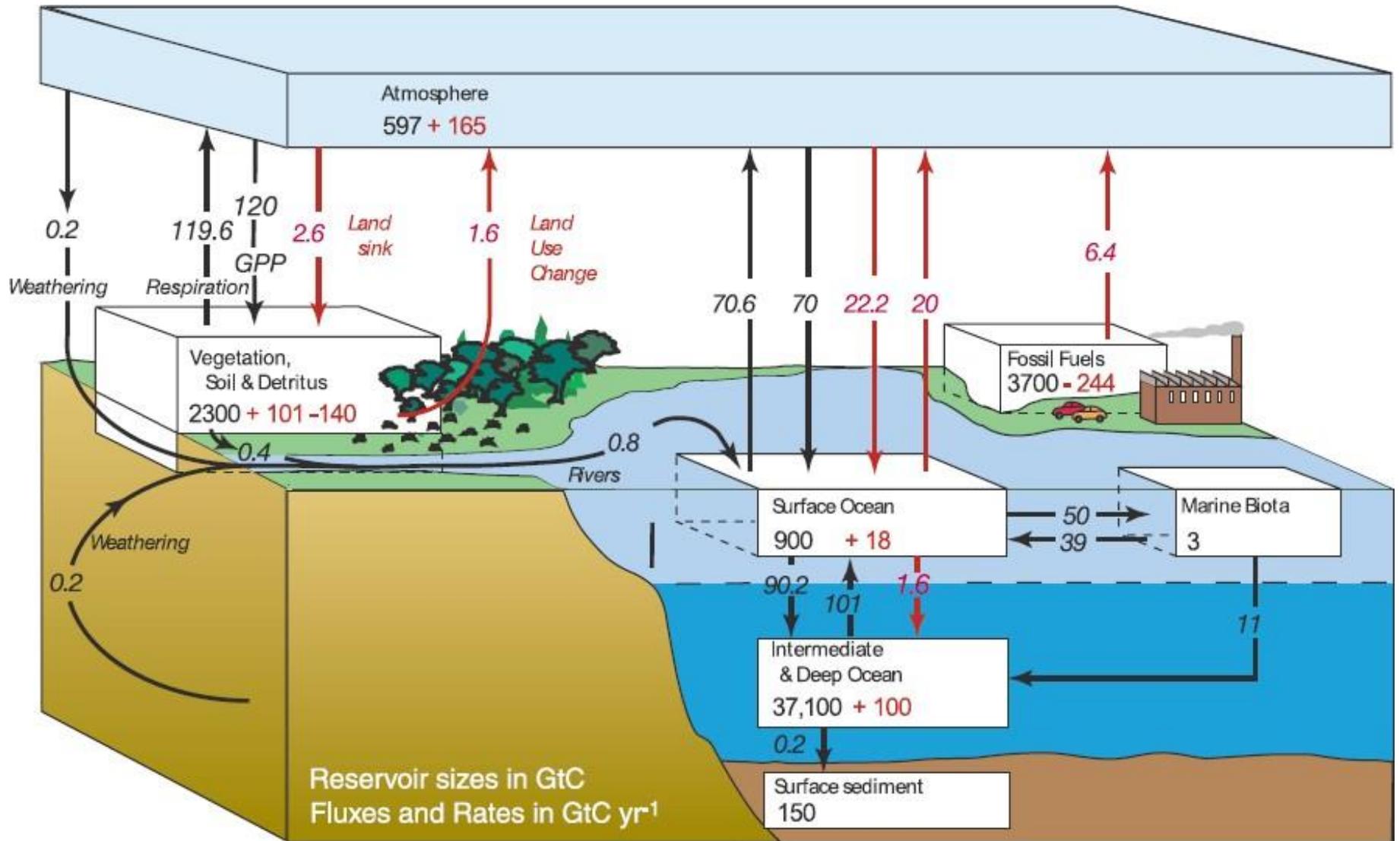
Large increases in greenhouse gases and aerosols since pre-industrial times

bl 5.1

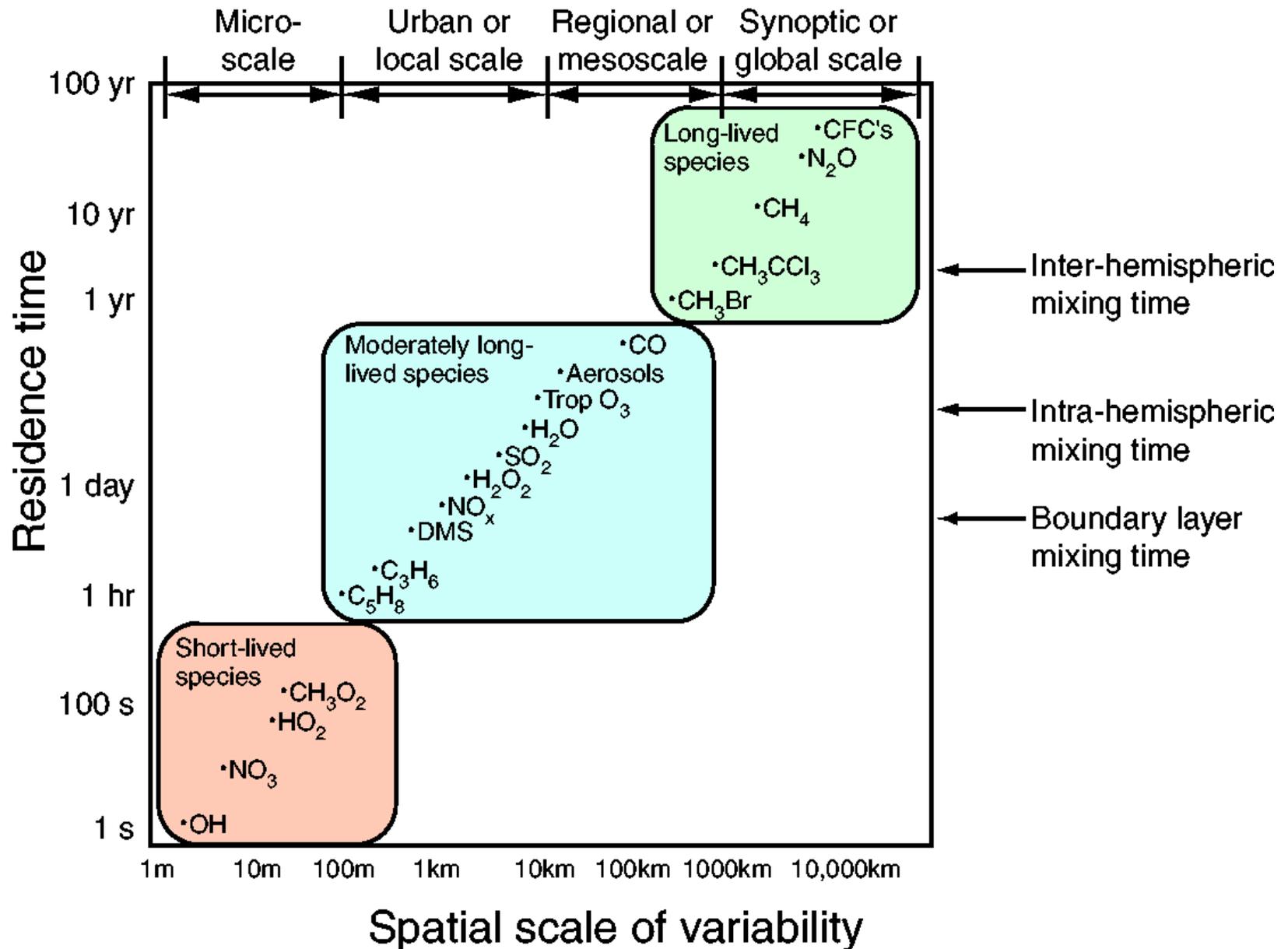
Some gases in dry tropospheric air^a

	Chemical formula	Fraction of volume of air occupied by the gas ^b	Residence time (or lifetime) ^c	Major sources	
	Nitrogen	N ₂	78.084%	1.6 × 10 ⁷ years	Biological
	Oxygen	O ₂	20.946%	3,000-4,000 years	Biological
	Argon	Ar	0.934%	—	Radiogenic
→	Carbon dioxide	CO ₂	380 ppmv	3-4 years ^d	Biological, oceanic combustion (Concentration increasing)
	Neon	Ne	18.18 ppmv		Volcanic (?)
	Helium	He	5.24 ppmv		Radiogenic
→	Methane	CH ₄	1.7 ppmv	9 years	Biological, anthropogenic
	Hydrogen	H ₂	0.56 ppmv	~2 years	Biological, anthropogenic
→	Nitrous oxide		0.31 ppmv	150 years	Biological, anthropogenic
→	Carbon monoxide		40-200 ppbv	~60 days	Photochemical, anthropogenic
→	Ozone	O ₃	10-500 ppbv	Hours	Photochemical
→	Non-methane hydrocarbons (NMHC)		5-20 ppbv	Variable	Biological, anthropogenic
→	Halocarbons		3-10 ppbv	Variable	Mainly anthropogenic
	Hydrogen peroxide	H ₂ O ₂	0.1-1 ppbv	1 day	Photochemical
	Formaldehyde	HCHO	0.1-1 ppbv	~1.5 hours	Photochemical
→	Nitrogen species (NO + NO ₂ (=NO _x) + NO ₃ + N ₂ O ₅ + HNO ₃ + PAN)	NO _y	0 pptv- 1 ppmv	Variable	Soils, anthropogenic, lightning
→	Ammonia	NH ₃	10 pptv-1 ppbv	1-10 days	Biological
→	Sulfur dioxide	SO ₂	10 pptv-1 ppbv	Days	Photochemical, volcanic, anthropogenic
	Dimethyl sulfide (DMS)	CH ₃ SCH ₃	10-100 pptv	0.7 days	Biological
	Hydrogen sulfide	H ₂ S	5-500 pptv	1-5 days	Biogenic, volcanic
→	Carbon disulfide	CS ₂	1-300 pptv	~120 hours	Biological, anthropogenic
	Hydroxyl radical	OH	0.1-10 pptv	~1 second	Photochemical
	Hydroperoxyl radical ^e	HO ₂	0.1-10 pptv		Photochemical

The global carbon cycle for the 1990s



Residence times and spatial variability

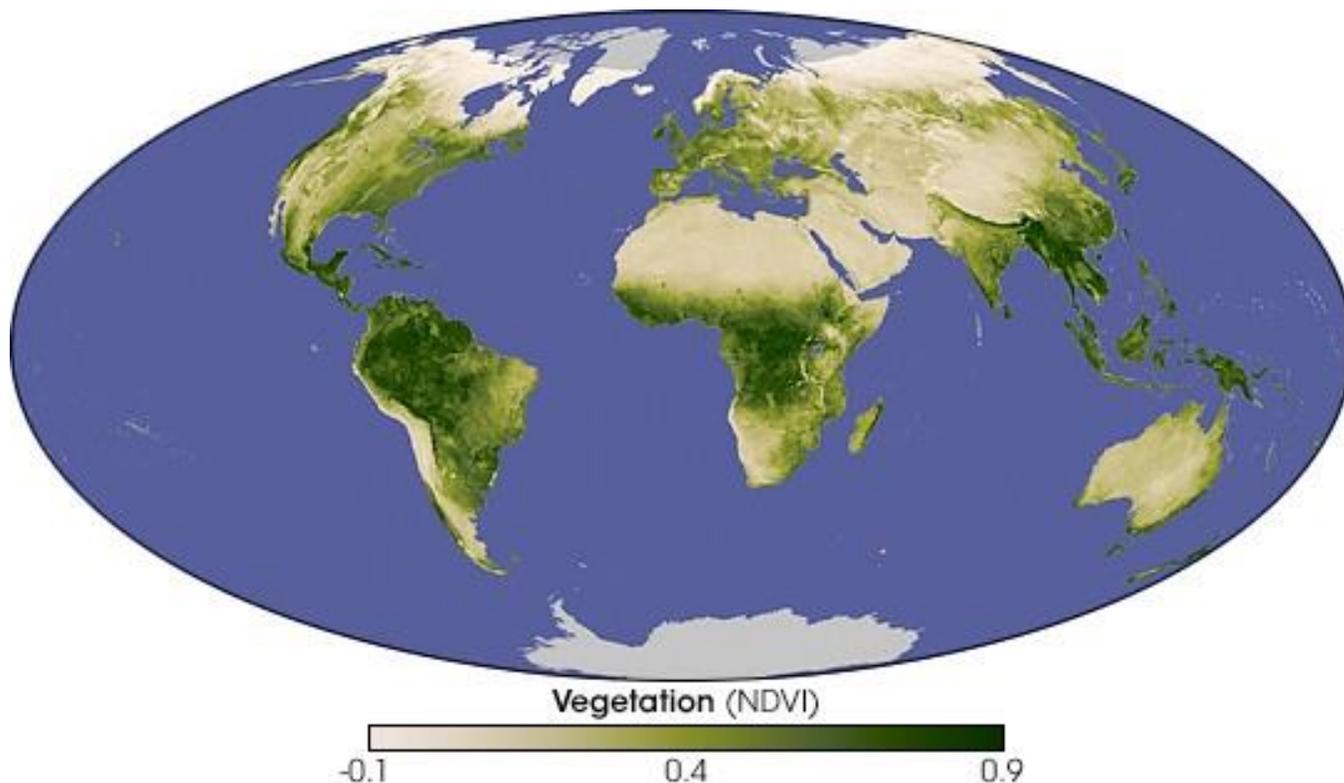


Sources of gases

- Biological: Land/Ocean
- Solid earth
- Mechanical generation (sea salt and dust)
- Anthropogenic: Fossil fuel, biomass burning, industrial
- Photochemical

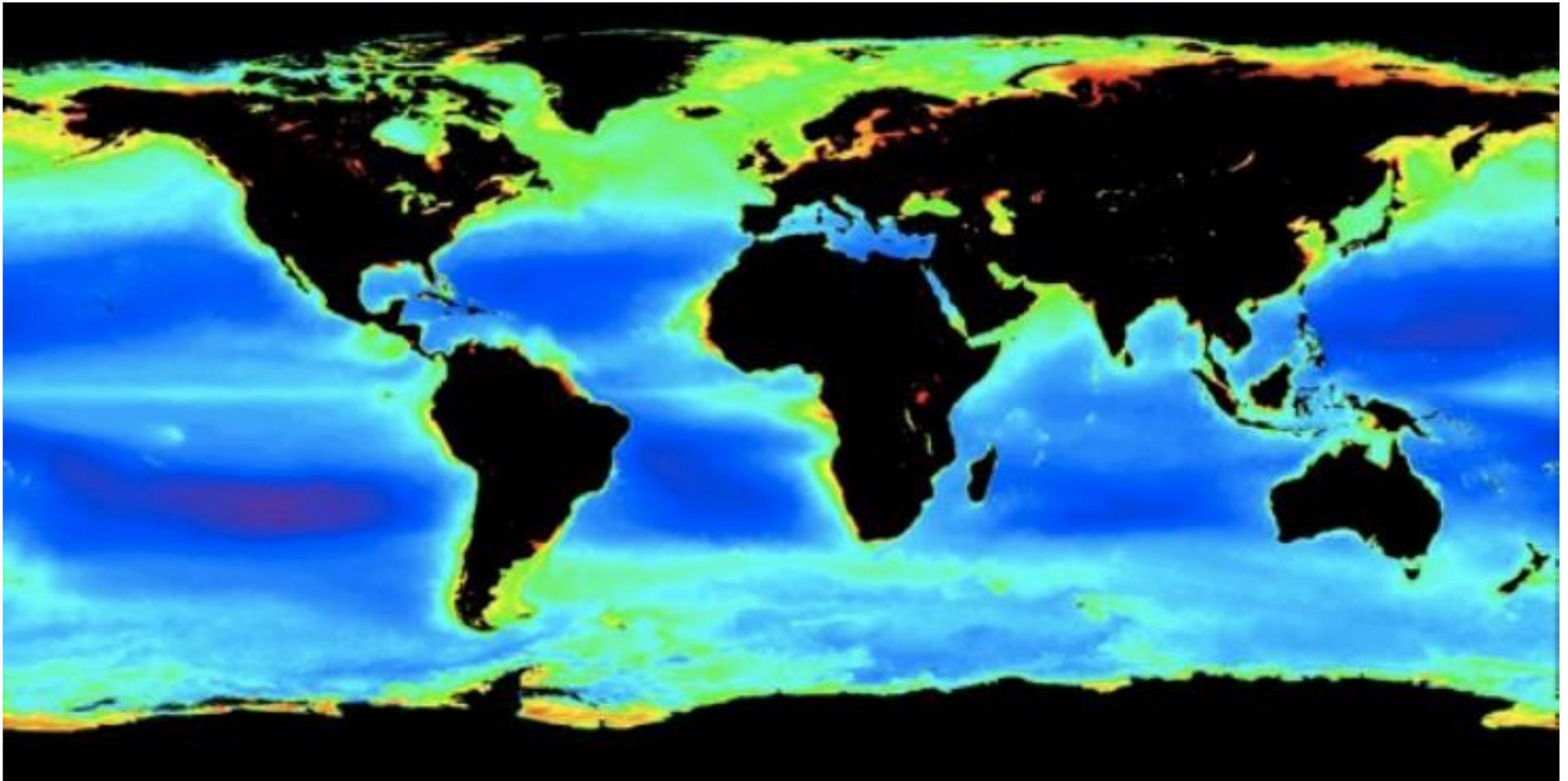
Biological sources - Land

- 
- Biological source of CH₄: swamps, rice paddies, termites, ruminants
- Biological sources of N₂O and NO: nitrifying and denitrifying bacteria in soils
- Plants: Reactive hydrocarbons (isoprene, terpenes, etc...)



Biological sources - Oceans

- Oceans: DMS, H₂S, COS, CH₃Cl, CH₃I, hydrocarbons



Solid earth



Figure 13. Eruption of Mount Pinatubo in the Phillipines in 1991. This type of explosive eruption injects large amounts of material into the stratosphere to altitudes of greater than 30 km.

- Volcanoes: SO_2 , H_2S , COS , particles, H_2O , CO_2 , HCl ...
- Rocks (radiogenic): He (decay of uranium and thorium), Argon (decay potassium-40), radon (decay of uranium-238)
- Weathering of CaCO_3 rocks

Dust and sea salt = generation by the action of wind

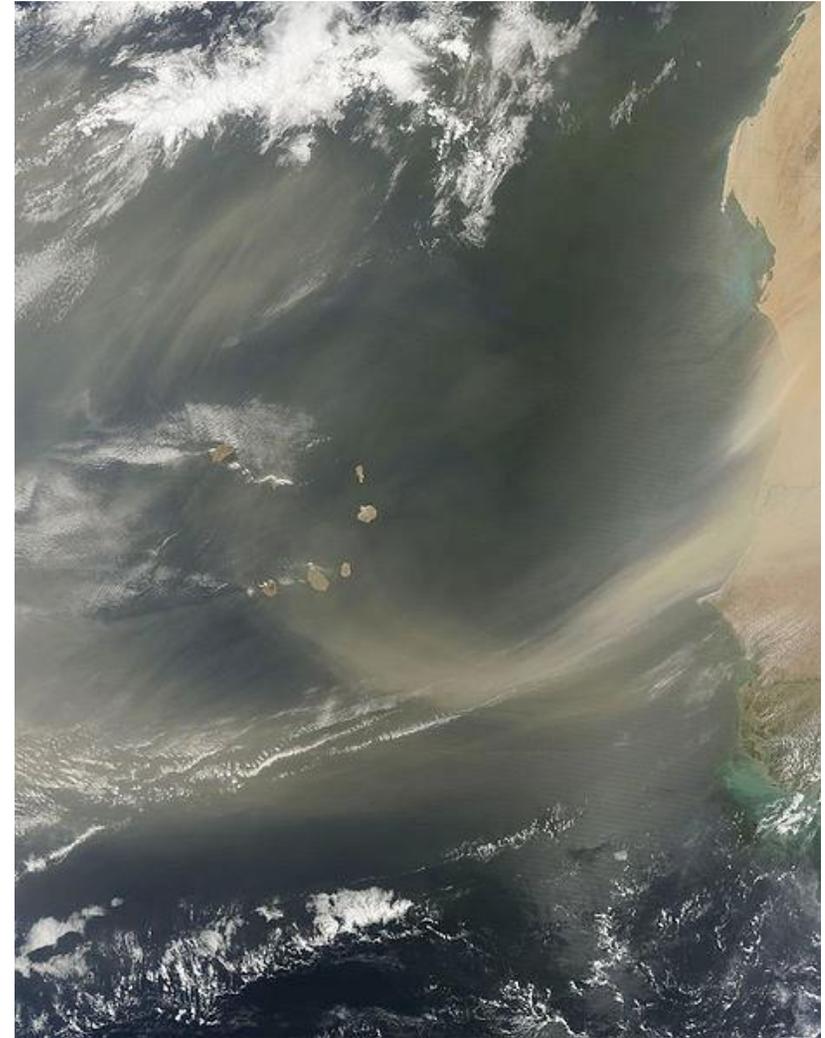
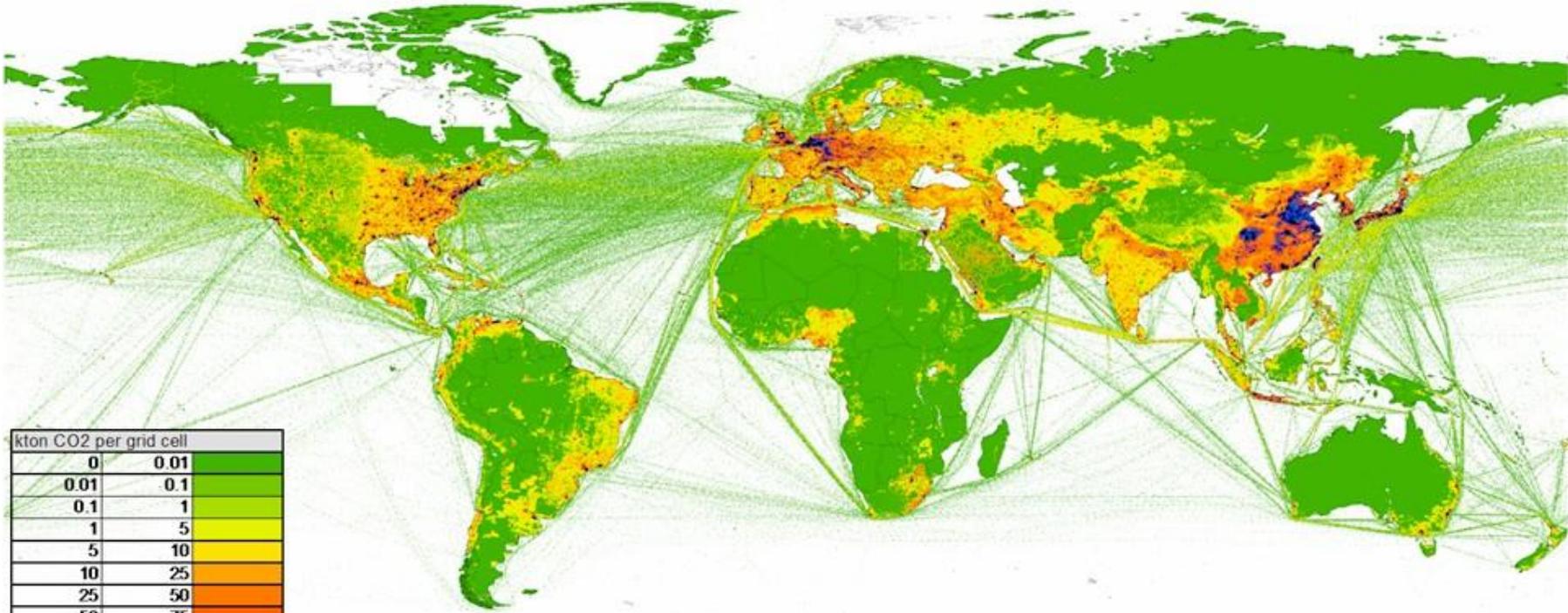


Figure 1. Air entrainment in sea water and aerosol injection into the atmosphere.

Fossil fuel emissions
(concentrated in the northern hemisphere)

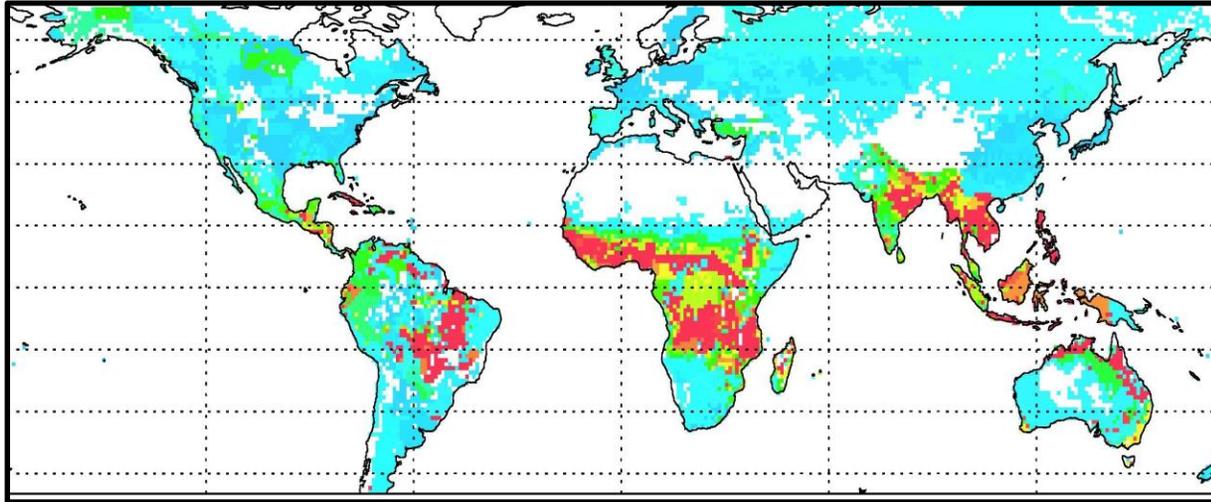


kton CO2 per grid cell	
0	0.01
0.01	0.1
0.1	1
1	5
5	10
10	25
25	50
50	75
75	100
100	150
150	200
200	

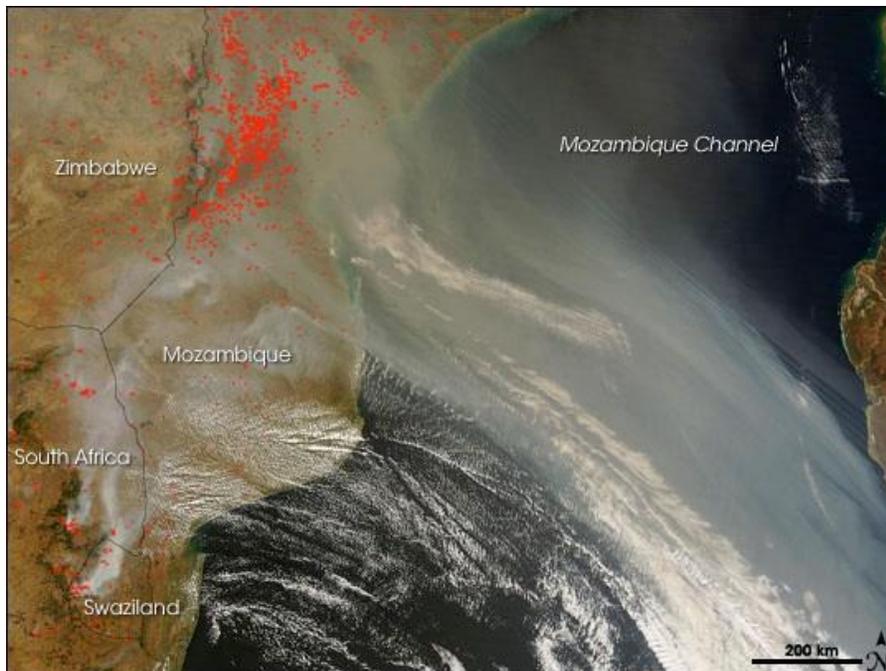
EC-JRC/PBL. EDGAR version 4.0. <http://edgar.jrc.ec.europa.eu/>, 2009



Biomass burning

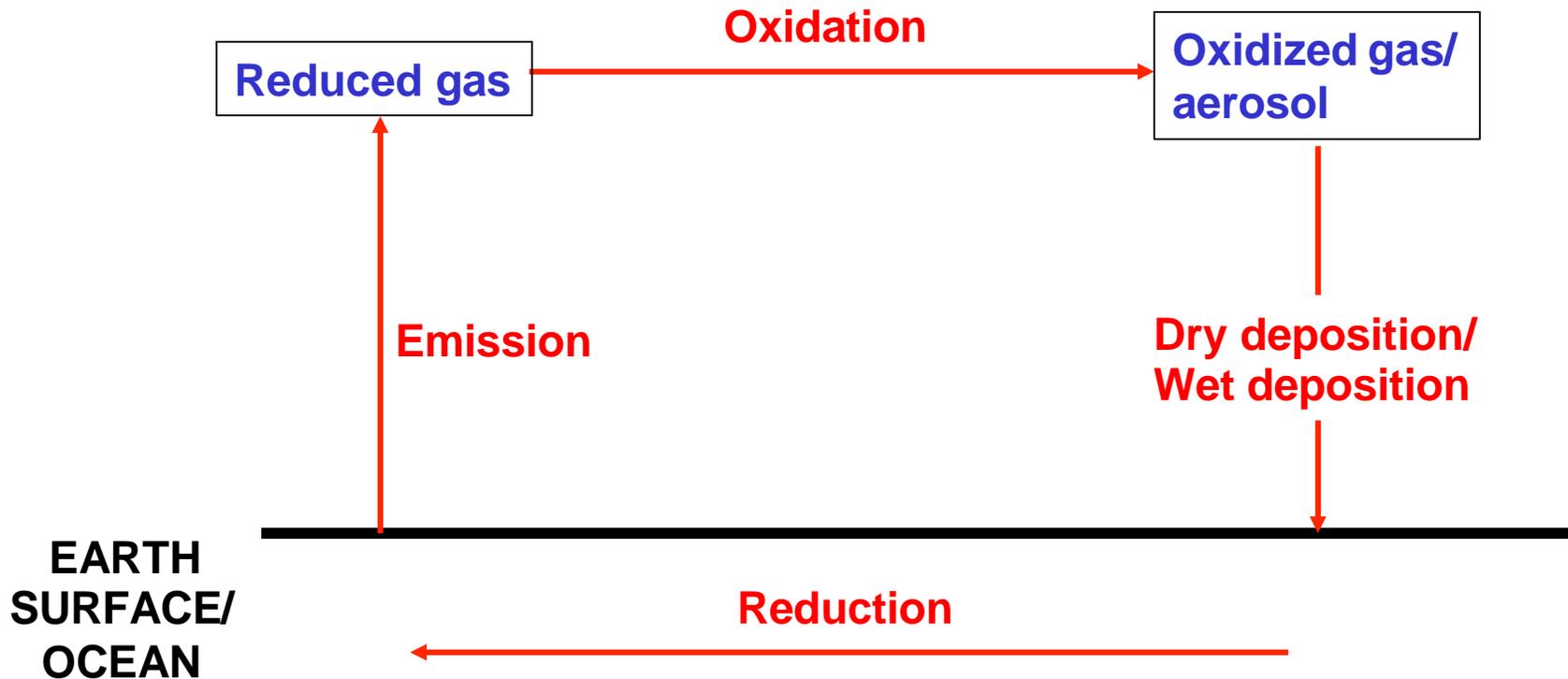


Biomass burning emissions
(concentrated in the tropics)
Agricultural and natural (lightning)



Photochemical sources/sinks

- Gases produced photochemically in the atmosphere: O_3 , OH , HO_2 , $HCHO$, H_2O_2 , CH_3OOH , NO_2 , NO_3 , HNO_3 , CO , SO_2 , etc...
- Sinks: atmospheric oxidation followed by dry or wet deposition



Some important Trace gases

- Hydroxyl radical (OH)
- Reactive nitrogen species and the nitrogen cycle
- Hydrocarbons, Carbon monoxide
- Ozone