Mid & Quize 3/5 — 40/100 — Fourty only

Physical_Chemistry_2nd_YUGS_EV_ST

Name of a student ---------- Signature ---------- No. ----------

Mustansiriyah University
Department of Chemistry

2nd SEM-2025_Bologna_Process
Mid_Exam_Class_B_Paper_A

Q1/ MCQ test (Answer the following) (Marks 50 %)

1: The reduced phase rule is interested in two variants?
Answer: a) p & T b) F & T c) p & conc. d) T & conc.

2: Ideal solution follows - - - - - - - - - - law.
Answer: a) Raoult's b) Trouton's c) Henry's law d) Van't Hoff's law

3: The three phases of H_2O in the phase diagram meets?
Answer: a) at 1 atm b) over 1 atm c) below 1atm d) at any pressure

4: Liquid solution of HNO_3 is formed from?
Answer: a) 1 C b) 2 C c) 3 C d) 4 C

5: How many phases are there when the number of variants is zero and the number of components is one?
Answer: a) zero b) 1 c) 2 d) 3

6: The Clausius-Clapeyron equation can be applied when there is an equilibrium between one of the following?
Answer: a) L & L b) S & L c) G & L d) S & S

7: One of the following formulas represents the right equation of Henry's law?
Answer: a) $P_A = \chi_A P^*_A$ b) $P_A > \chi_A P^*_A$ c) $P_A < \chi_A P^*_A$ d) none of these

8: Molality is used to calculate the molar mass of the?
Answer: a) non-volatile solute b) pure solute c) pure solvent d) solution

9: Osmosis pressure exerts when the solvent transfers to the?
Answer: a) volatile solute b) non-volatile solute c) pure solvent d) solution

10- One of the most important benefits of measuring ΔVP, ΔT_b, ΔT_f and $\Delta\Pi$ is to calculate - - - - - of B?
Answer: a) M b) m c) V d) p

Q2/ The vapor pressure (VP) of a substance is 30 torr at 250 K. At what temperature will the substance have VP of 150 torr? $\Delta_{vap}H$ is 45 kJ mol^{-1}? (Marks 25%)

Q3/ Plot the phase diagram of the system (A & B) assumed that (A & B) do not react with each other. A freezes at (− 5 °C) and B freezes at (7 °C), and that an eutectic mixture is formed when the ratio is 70 wt % of A and that the eutectic melts at (− 10°C), then label all the parts (p & F) of the diagram? (Marks 25%)

A = 70%
B = 30%

Thur_13-03-2025 — Best wishes — Dr Abduljabbar I. R. Rushdi

Q2) $P_i = 30$ torr / $P_f = 150$ torr / $T_i = 250K$ / $T_f = ?$

$\Delta_{vap}H = 45\ J K mol^{-1}$.

$$\ln \frac{P_f}{P_i} = -\frac{\Delta_{vap}H}{R}\left(\frac{1}{T_f} - \frac{1}{T_i}\right)$$

Q2 15/25

$$\ln \frac{150 torr}{30 torr} = -\frac{45\ J K mol^{-1}}{8.314\ J K mol^{-1}}\left(\frac{1}{T_F} - \frac{1}{250K}\right)$$

$$\ln 5 = -5.4\left(\frac{1}{T_F} - \frac{1}{250K}\right)$$

54112.6

$$1.6 = -5.4\left(\frac{1}{T_F} - \frac{1}{250K}\right)\Big] \div (-5.4)$$

54112.6

$$-3.4 = \left(\frac{1}{T_F} - \frac{1}{250K}\right) \Rightarrow -3.4 = (T_F - 250K)$$

$$-3.4 + 250K = T_F \Longrightarrow \boxed{246.6\ K}$$

Q3 Zero/25