



1/5 Mid P Quize

Physical\_Chemistry\_2<sup>nd</sup>\_YUGS\_EV\_ST

20/100 Twenty only



Name of a student ----- Signature ----- No. -----

Mustansiriyah University  
Department of Chemistry

2<sup>nd</sup> SEM-2025 Bologna Process  
Mid Exam Class\_B\_Paper\_A

Q1/ MCQ test (Answer the following)

(Marks 50 %)

1: The reduced phase rule is interested in two variants?

- Answer:  a) p & T  b) F & T  c) p & conc.  d) T & conc.

2: Ideal solution follows ----- law.

- Answer:  a) Raoult's  b) Trouton's  c) Henry's law  d) Van't Hoff's law

3: The three phases of H<sub>2</sub>O in the phase diagram meets?

- Answer:  a) at 1 atm  b) over 1 atm  c) below 1atm  d) at any pressure

4: Liquid solution of HNO<sub>3</sub> is formed from?

- Answer:  a) 1 C  b) 2 C  c) 3 C  d) 4 C

5: How many phases are there when the number of variants is zero and the number of components is one?

- Answer:  a) zero  b) 1  c) 2  d) 3

6: The Clausius-Clapeyron equation can be applied when there is an equilibrium between one of the following?

- Answer:  a) L & L  b) S & L  d) G & L  d) S & S

7: One of the following formulas represents the right equation of Henry's law?

- Answer:  a)  $P_A = \chi_A P^*_{A}$   b)  $P_A > \chi_A P^*_{A}$   c)  $P_A < \chi_A P^*_{A}$   d) none of these

8: Molality is used to calculate the molar mass of the?

- Answer:  a) non-volatile solute  b) pure solute  c) pure solvent  d) solution

9: Osmosis pressure exerts when the solvent transfers to the?

- Answer:  a) volatile solute  b) non-volatile solute  c) pure solvent  d) solution

10- One of the most important benefits of measuring  $\Delta VP$ ,  $\Delta T_b$ ,  $\Delta T_f$  and  $\Delta H$  is to calculate ----- of B?

- Answer:  a) M  b) m  c) V  d) p

Q2/ The vapor pressure (VP) of a substance is 30 torr at 250 K. At what temperature will the substance have

VP of 150 torr?  $\Delta_{vap}H$  is 45 kJ mol<sup>-1</sup>?

(Marks 25%)

Q3/ Plot the phase diagram of the system (A & B) assumed that (A & B) do not react with each other. A

freezes at (-5 °C) and B freezes at (7 °C), and that an eutectic mixture is formed when the ratio is 70 wt % of A and that the eutectic melts at (-10°C), then label all the parts (p & F) of the diagram? (Marks 25%)

$$P_i = 30 \text{ torr}$$

$$T_i = 250 \text{ K}$$

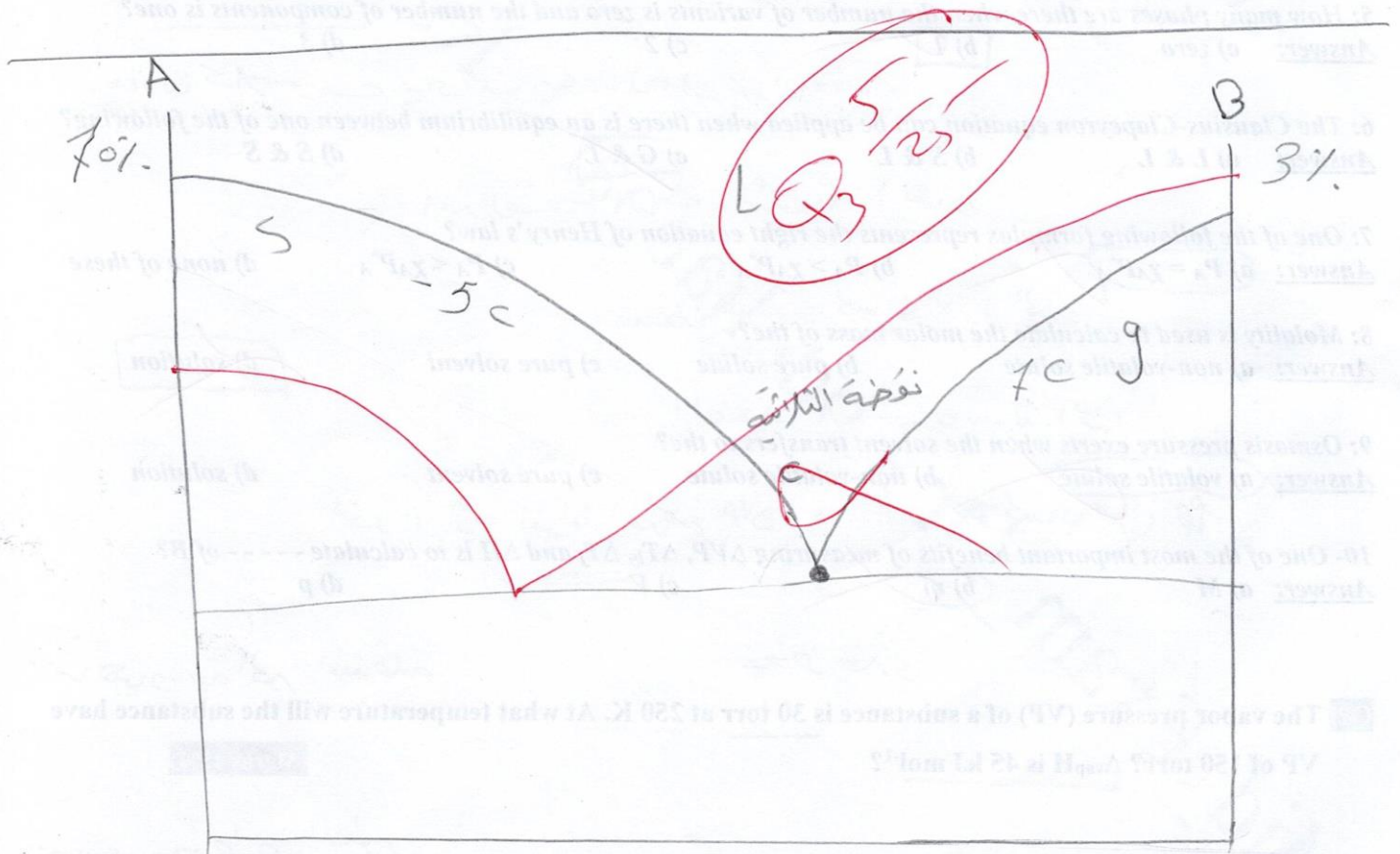
$$\Delta_{\text{vap}}H = 45 \text{ kJ mol}^{-1}$$

$$P_f = 150 \text{ torr}$$

$$T_f = ?$$

$$\ln \frac{P_i}{P_f} = \frac{\Delta_{\text{vap}}H}{R} \left( \frac{1}{T_i} - \frac{1}{T_f} \right)$$

~~Q2~~



~~Q3~~

~~$$F = C - P + 2$$~~

قاعدة الطور

$$F = C - P + 1$$

قاعدة الطور المتزن

التركيب 50%

التركيب 50%

القاعدة المتطلبية هنا